

	QMRF identifier (JRC Inventory): Q13-21c-0033
	QMRF Title: QSAR for abiotic degradation in water
	Printing Date: Dec 11, 2019

1. QSAR identifier

1.1. QSAR identifier (title):

QSAR for abiotic degradation in water

1.2. Other related models:

1.3. Software coding the model:

QSARModel 4.0.4

Molcode Ltd., Turu 2, Tartu, 51014, Estonia

<http://www.molcode.com>

2. General information

2.1. Date of QMRF:

05.05.2010

2.2. QMRF author(s) and contact details:

Molcode model development team Molcode Ltd. Turu 2, Tartu, 51014, Estonia

models@molcode.com <http://www.molcode.com>

2.3. Date of QMRF update(s):

2.4. QMRF update(s):

2.5. Model developer(s) and contact details:

Molcode model development team Molcode Ltd. Turu 2, Tartu, 51014, Estonia

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2.6. Date of model development and/or publication:

04.05.2010

2.7. Reference(s) to main scientific papers and/or software package:

[1]Karelson M, Dobchev D, Tamm T, Tulp I, Jänes J, Tämm K, Lomaka A, Savchenko D & Karelson G (2008). Correlation of blood-brain penetration and human serum albumin binding with theoretical descriptors. ARKIVOC 16, 38-60.

[2]Karelson M, Karelson G, Tamm T, Tulp I, Jänes J, Tämm K, Lomaka A, Savchenko D & Dobchev D (2009). QSAR study of pharmacological permeabilities. ARKIVOC 2, 218–238.

2.8. Availability of information about the model:

Training, selection and test sets are available. Algorithm is defined.

2.9. Availability of another QMRF for exactly the same model:

None to date.

3. Defining the endpoint - OECD Principle 1

3.1. Species:

3.2. Endpoint:

2.Environmental fate parameters 2.1.c.Persistence: Abiotic degradation in water. Other

3.3. Comment on endpoint:

The half-life is the time required for the concentration of a substance to halve its original value in a particular environmental medium. The half-lives of organic compounds are among the most commonly used

criteria for studying persistence [sect.9.2; ref 1]. The semi-quantitative data based on expert judgment and actual experimental values have already been suggested by Webster et al. [sect.9.2; ref 2] as preferable for half life identification, and are commonly used to develop the widely applied multimedia models [sect.9.2; ref 3,4]. In addition, a simple QSPR regression model has been demonstrated to be a useful tool for the identification and prioritization of existing or not yet synthesized potential persistent organic pollutants [sect.9.2; ref 5].

3.4.Endpoint units:

The half-life values (in h) were transformed into logarithmic form for modelling

3.5.Dependent variable:

$\log T(0.5)$

3.6.Experimental protocol:

3.7.Endpoint data quality and variability:

The dataset includes 250 organic compounds of known half-lives for transformation in air [sect.9.2; ref 6]. A collection of experimental data from different labs has been used. Semi-quantitative degradation half lives in water were organized in seven half-life categories.

Statistics: max value: 4.74; min value: 1.23; standard deviation: 0.770; skewness: 0.959.

4.Defining the algorithm - OECD Principle 2

4.1.Type of model:

QSAR

4.2.Explicit algorithm:

Multilinear regression QSAR

Multilinear regression QSAR derived by the BMLR (Best Multiple Linear Regression) method

$\log T(0.5) = 4.189$

$+0.104 \cdot \text{Kier\&Hall index (order 0)}$

$-2.974 \cdot \text{ZX Shadow / ZX Rectangle (AM1)}$

$+105.902 \cdot \text{HACA-2/TMSA (Zefirov) (all)}$

$-1.918 \cdot \text{Polarity parameter (Zefirov) / distance}$

$-0.662 \cdot \text{Square root of Charged (Zefirov) Surface Area of O atoms}$

$+0.999 \cdot \text{Min net atomic charge (AM1) for N atoms}$

4.3.Descriptors in the model:

[1]Kier&Hall index (order 0) unitless zero order Kier and Hall valence connectivity index

[2]ZX Shadow / ZX Rectangle (AM1) unitless Relative shadow area of a molecule in ZX plane

[3]HACA-2/TMSA (Zefirov) (all $\text{au}/\text{\AA}^2$ Area-weighted surface charge of hydrogen bonding acceptor atoms

4.4.Descriptor selection:

Initial pool of ~1000 descriptors. Stepwise descriptor selection based on a set of statistical selection rules:

a) one-parameter equations: Fisher criterion and R^2 over threshold, variance and t-test value over threshold, intercorrelation with another

descriptor not over threshold.

b) two-parameter equations: intercorrelation coefficient below threshold, significant correlation with endpoint, in terms of correlation coefficient and t-test.

Stepwise trial of additional descriptors not significantly correlated to any already in the model.

4.5. Algorithm and descriptor generation:

1D, 2D, and 3D theoretical calculations. Quantum chemical descriptors derived from AM1 calculation. Model developed by using multilinear regression.

4.6. Software name and version for descriptor generation:

QSARModel 4.0.4

QSAR/QSPR package that will compute chemically meaningful descriptors and includes statistical tools for regression modeling

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4.7. Chemicals/Descriptors ratio:

27.7 (166 chemicals / 6 descriptors)

5. Defining the applicability domain - OECD Principle 3

5.1. Description of the applicability domain of the model:

Applicability domain based on training set:

- a) by chemical identity: diverse set of organic pollutants (aromatic, aliphatic and cyclic amines, ketones, alcohols, esters, etc)
- b) by descriptor value range: The model is suitable for compounds that have the descriptors in the following minimal-maximal ranges:

Kier&Hall index (order 0): 1.43 - 15.5

ZX Shadow / ZX Rectangle (AM1): 0.552 - 0.857

HACA-2/TMSA (Zefirov) (all): 0.00 - 0.0107

Polarity parameter (Zefirov) / distance: 0.00561 - 0.273

Square root of Charged (Zefirov) Surface Area of O atoms: 0.00 - 2.91

Min net atomic charge (AM1) for N atoms: -0.443 - 0.578

5.2. Method used to assess the applicability domain:

Range of descriptor values in training set with $\pm 30\%$ confidence.

Descriptor values must fall between maximal and minimal descriptor values of training set $\pm 30\%$.

5.3. Software name and version for applicability domain assessment:

QSARModel 4.0.4

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5.4. Limits of applicability:

6.Internal validation - OECD Principle 4

6.1.Availability of the training set:

Yes

6.2.Available information for the training set:

CAS RN: Yes

Chemical Name: Yes

Smiles: No

Formula: Yes

INChI: No

MOL file: Yes

6.3.Data for each descriptor variable for the training set:

All

6.4.Data for the dependent variable for the training set:

All

6.5.Other information about the training set:

During the modeling procedure one compound was excluded as a statistical outlier (2,2',3,3',4,5,5',6,6'-nonachlorobiphenyl).

The final training set consisted of 166 data points: 0 negative values; 166 positive values.

6.6.Pre-processing of data before modelling:

$R^2 = 0.715$ (Correlation coefficient)

$s_2 = 0.420$ (Standard error of the estimate)

$F = 80.3$ (Fisher function)

6.7.Statistics for goodness-of-fit:

6.8.Robustness - Statistics obtained by leave-one-out cross-validation:

$R^2_{CV} = 0.689$

6.9.Robustness - Statistics obtained by leave-many-out cross-validation:

$R^2_{CVMO} = 0.685$

6.10.Robustness - Statistics obtained by Y-scrambling:

6.11.Robustness - Statistics obtained by bootstrap:

6.12.Robustness - Statistics obtained by other methods:

ABC analysis (2:1 training : prediction) on sorted (in increasing order of endpoint value) data divided into 3 subsets (A;B;C). Training set formed with 2/3 of the compounds (set A+B, A+C, B+C) and validation set consisted of 1/3 of the compounds (C, B, A).

average $R^2(\text{fitting}) = 0.719$

average $R^2(\text{prediction}) = 0.687$

7.External validation - OECD Principle 4

7.1.Availability of the external validation set:

Yes

7.2.Available information for the external validation set:

CAS RN: Yes

Chemical Name: Yes

Smiles: No
Formula: Yes
INChI: No
MOL file: Yes

7.3.Data for each descriptor variable for the external validation set:

All

7.4.Data for the dependent variable for the external validation set:

All

7.5.Other information about the external validation set:

83 data points: 0 negative values; 83 positive values

7.6.Experimental design of test set:

From sorted source data, each 3rd was subjected to the test set.

7.7.Predictivity - Statistics obtained by external validation:

$R^2 = 0.822$ (Coefficient of determination)

7.8.Predictivity - Assessment of the external validation set:

All are in range of applicability domain:

Kier&Hall index (order 0): 1.45 - 17.3

ZX Shadow / ZX Rectangle (AM1): 0.566 - 0.800

HACA-2/TMSA (Zefirov) (all): 0.00 - 0.0108

Polarity parameter (Zefirov) / distance: 0.00626 - 0.273

Square root of Charged (Zefirov) Surface Area of O atoms: 0.275 - 2.45

Min net atomic charge (AM1) for N atoms: -0.395 - 0.580

7.9.Comments on the external validation of the model:

The validation coefficient of determination (R^2) is even better than the coefficient of the model based on the training set.

8.Providing a mechanistic interpretation - OECD Principle 5

8.1.Mechanistic basis of the model:

The model comprises two size and shape related descriptors, particularly "Kier&Hall index (order 0)" and "ZX Shadow / ZX Rectangle (AM1)". Hydrogen bonding capabilities are taken into account with the descriptor "HACA-2/TMSA (Zefirov) (all)". Hydrogen bonds tend to stabilize the structure and therefore suppress the degradation. In contrast, "Polarity parameter (Zefirov) / distance" represents structure inner strains and increases the degradation. The other two descriptors "Square root of Charged (Zefirov) Surface Area of O atoms" and "Min net atomic charge (AM1) for N atoms" are atom specific and reflect oxygen and nitrogen contributions to the degradation, respectively.

8.2.A priori or a posteriori mechanistic interpretation:

A posteriori mechanistic interpretation.

8.3.Other information about the mechanistic interpretation:

A similar interpretation can be found in the scientific literature [sect 9.2, ref 5]

9.Miscellaneous information

9.1. Comments:

The data were gathered from the Physical-Chemical Properties and Environmental Fate Handbook [sect.9.2; ref.6], which includes data from different sources. Therefore the experimental protocol cannot be provided. The data were also semi-quantitatively classified as proposed by Webster et al [sect.9.2: ref.2].

9.2. Bibliography:

- [1]UNEP, Stockholm Convention on Persistent Organic Pollutants, United Nations Environment Program, Geneva, Switzerland, 2000 <http://www.pops.int>
- [2]Webster E, Mackay D & Wania F (1998). Evaluating Environmental Persistence. Environmental Toxicological Chemistry 17, 2148–2158.
- [3]Klasmeier J, Matthies M, MacLeod M, Fenner K, Scheringer M, Stroebe M, Le Gall A C, McKone T, Van De Meent D & Wania F (2006). Application of Multimedia Models for Screening Assessment of Long-Range Transport Potential and Overall Persistence. Environmental Science & Technology 40, 53-60.
- [4]Fenner K, Scheringer M, Macleod M, Matthies M, McKone T, Stroebe M, Beyer A, Bonnell M, Le Gall A C, Klasmeier J, Mackay D, Van de Meent D, Pennington D, Scharenberg B, Suzuki N & Wania F (2005). Comparing Estimates of Persistence And Long-Range Transport Potential among Multimedia Models. Environmental Science and Technology 39, 1932-1942.
- [5]Gramatica P & Papa E (2007). Screening and ranking of POPs for global half-life: QSAR approaches for prioritization based on molecular structure. Environmental Science and Technology 41, 2833–2839.
- [6]Mackay D, Shiu WY & Ma KC (2000). Physical-Chemical Properties and Environmental Fate Handbook. CRCnet-BASE CD-ROM. Chapman and Hall/CRC, Boca Raton, FL, USA.

9.3. Supporting information:

Persistence water training_266.sdf	http://qsar.db.jrc.ec.europa.eu/qmrf/protocol/Q13-21c-0033/attachment/A692
Persistence water test_83.sdf	http://qsar.db.jrc.ec.europa.eu/qmrf/protocol/Q13-21c-0033/attachment/A693

Test set(s)

10. Summary (JRC QSAR Model Database)

10.1. QMRF number:

Q13-21c-0033

10.2. Publication date:

2013-06-27

10.3. Keywords:

Molcode; persistence; abiotic degradation; water; biodegradation;

10.4. Comments:

former Q8-10-30-265